

Hydraulic Interconnection between the Volcanic Aquifer and Springs, Lake Tana Basin on the Upper Blue Nile

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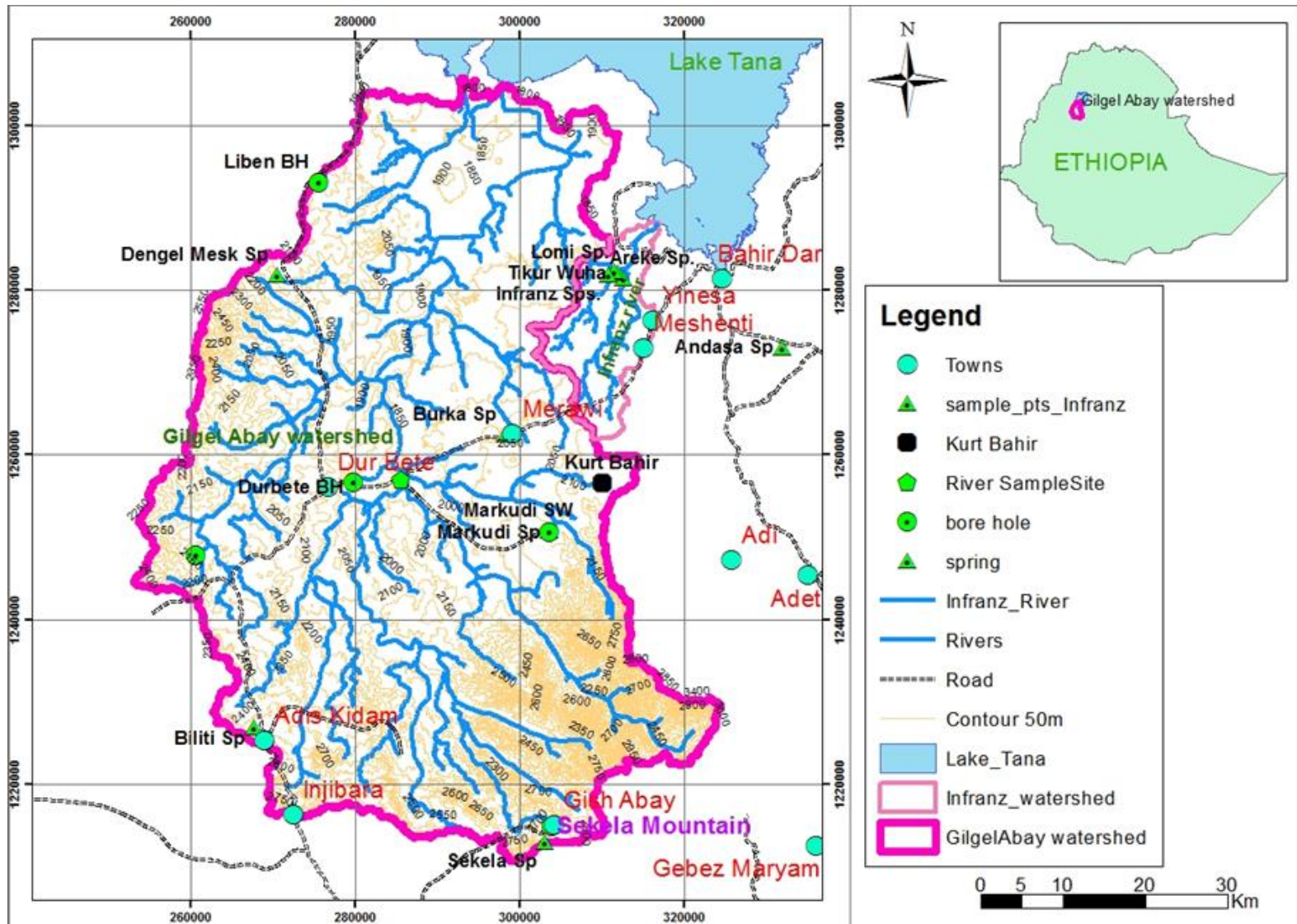


N°abstract

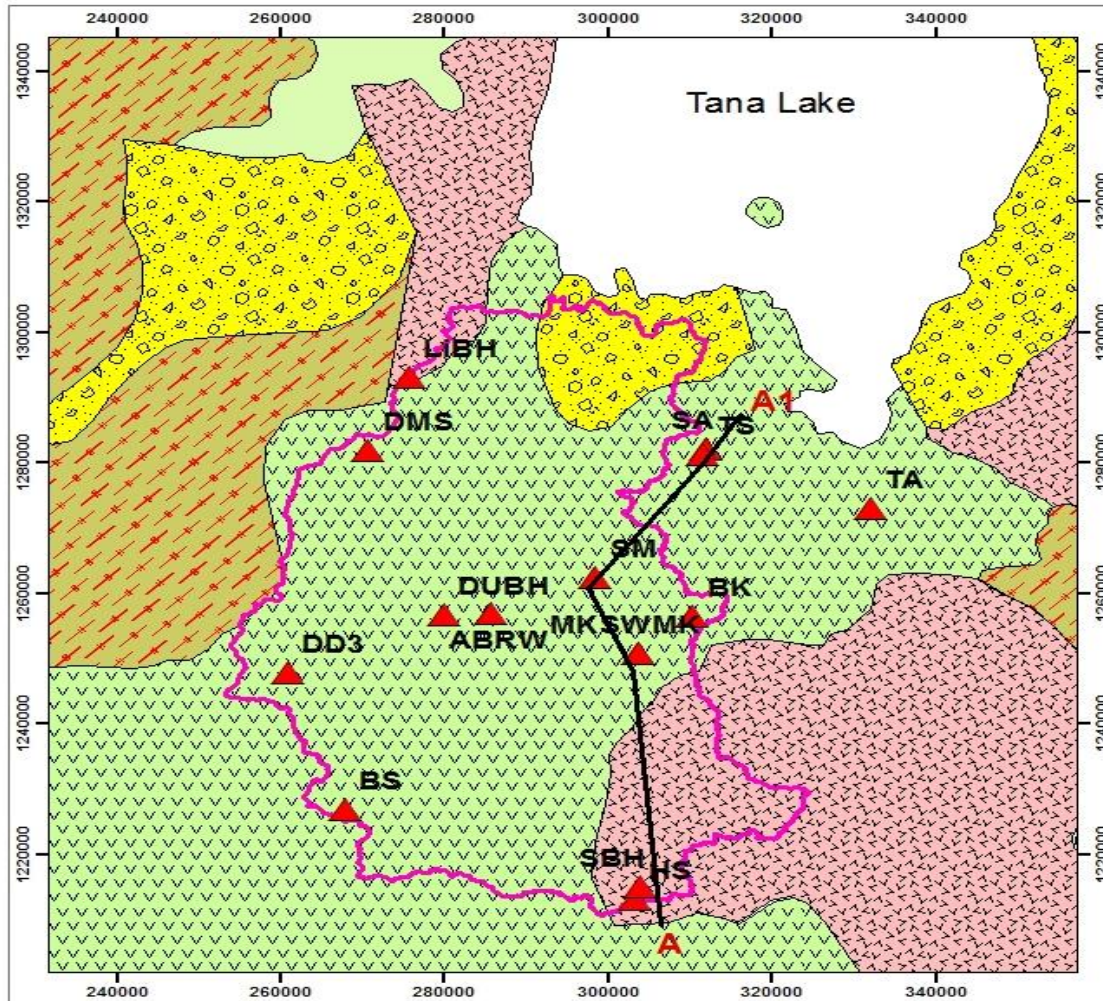


Introduction




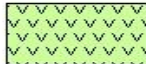



- Location of the study area

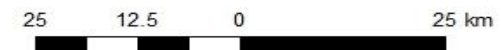


Geology and hydrogeology



Legend

-  Water sample points
-  Gilgel Abay watershed
-  Alluvial Deposit
-  Quaternary Basalt
-  Tarmaber Formation: Trap Series Basalt
-  Ashangi Formation: Trap Series Basalt
-  A — A1 Cross Section



Geological outcrops at the spring area



Recent Quaternary basalt: Highly vesicular, scoraceous and blocky nature and forms small hills here and there

Major springs in the study area

Name of Major springs	Estimated discharge	characteristics	Extraction to water supply
Areke Spring	estimated discharge > 165 l/s; some technical reports and thesis estimated differently, 210 l/s	They are highly diffused to estimate correctly	51 % of estimated discharge
Lomi spring	estimated discharge > 58 l/s, some technical reports estimated 90 l/s		41 % of estimated discharge
Infranz springs (two twin springs)	Estimated > 60 l/s		Not known
Tikur Wuha spring	estimated discharge > 30 l/s		
Merawi spring	Not estimated		Not known
Andasa spring	Not estimated		Not used



Problem statement

- The high discharge springs, namely: Areke, Lomi, Infranz, Tikur Wuha and others around Bahir Dar; Andasa (Tikurit) spring, and Merawi (Burka) springs are not well studied, and
 - it remains unclear whether they are regionally or locally sourced, because of the limited set of physical, chemical isotopic characteristics.

Objectives of the Study

- The study was conducted:
 - ❖ to identify the hydrological interconnection between these springs with the surrounding regional and local volcanic aquifer systems

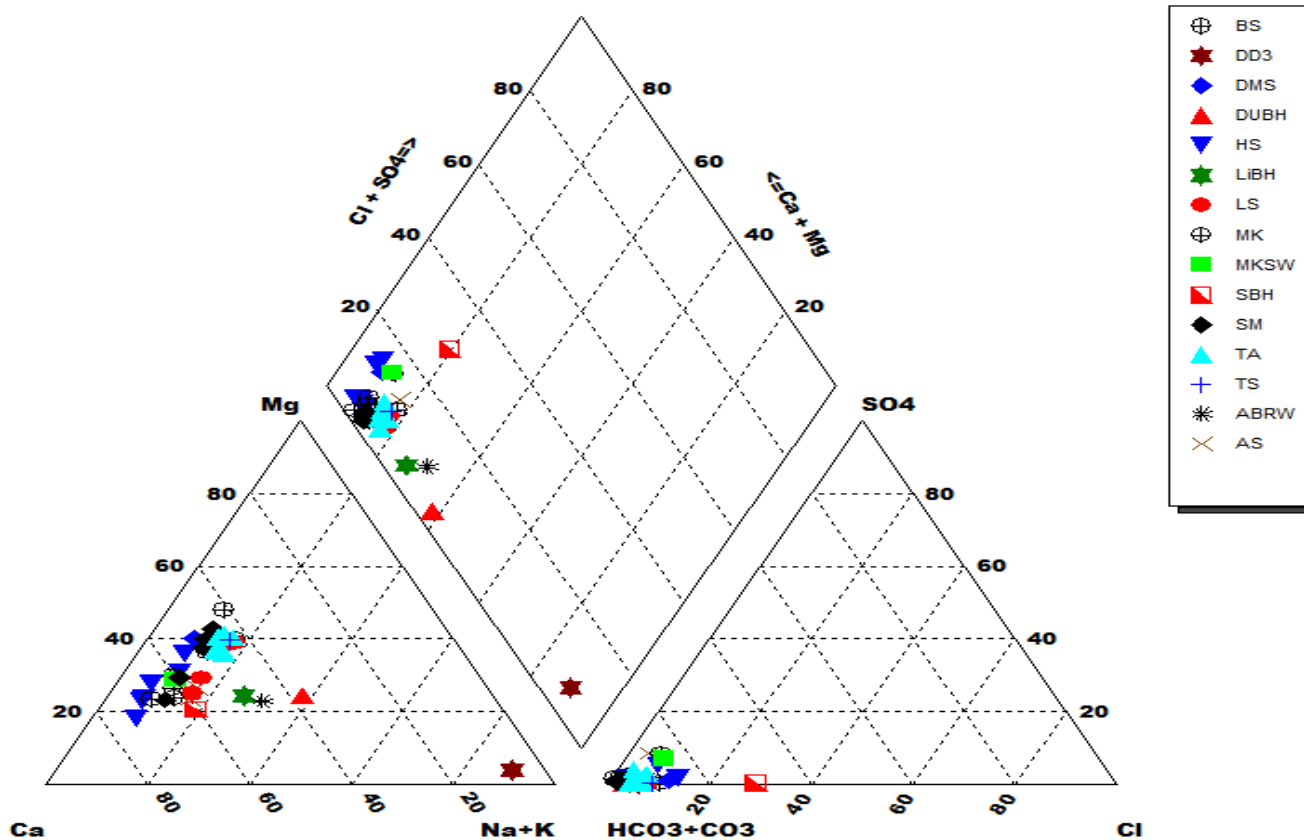
Methodology and Approach

Water sampling for chemistry and isotope analysis

- 1 spring at highlands of the area,
- 1 spring in transition zone and
- 3 springs in the lowlands of the study area were selected for **continuous** physico-chemical and isotope data collection and analysis for one year, June 2012-June 2013.
- 1 time snapshot samples were collected from boreholes, surface water and other available springs.

- The sampling sites were selected based on
 - ❖ the accessibility,
 - ❖ groundwater flow direction and
 - ❖ physiographic distribution in the catchment
- The collected water samples were analyzed for cations and anions; and stable isotopes

Result and Discussion

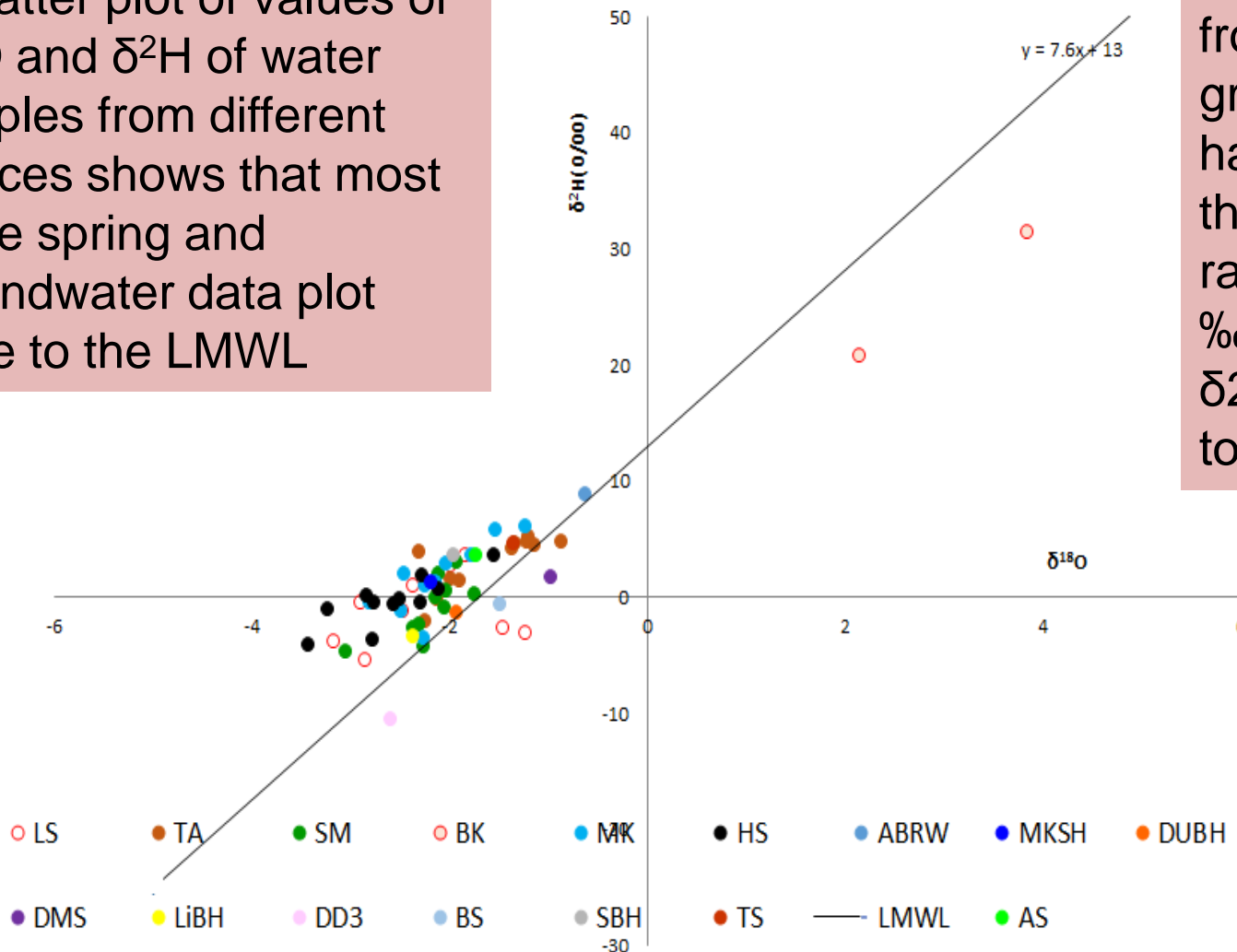


- Samples from the springs and wells have low TDS (less than 400 mg/l) and low EC (less than 400 μ S/cm).
- Major ions collected from springs, wells and surface water are plotted in the piper diagram . The result shows that most of the waters are very similar and fall in the same cluster of Ca-HCO₃ type, except some

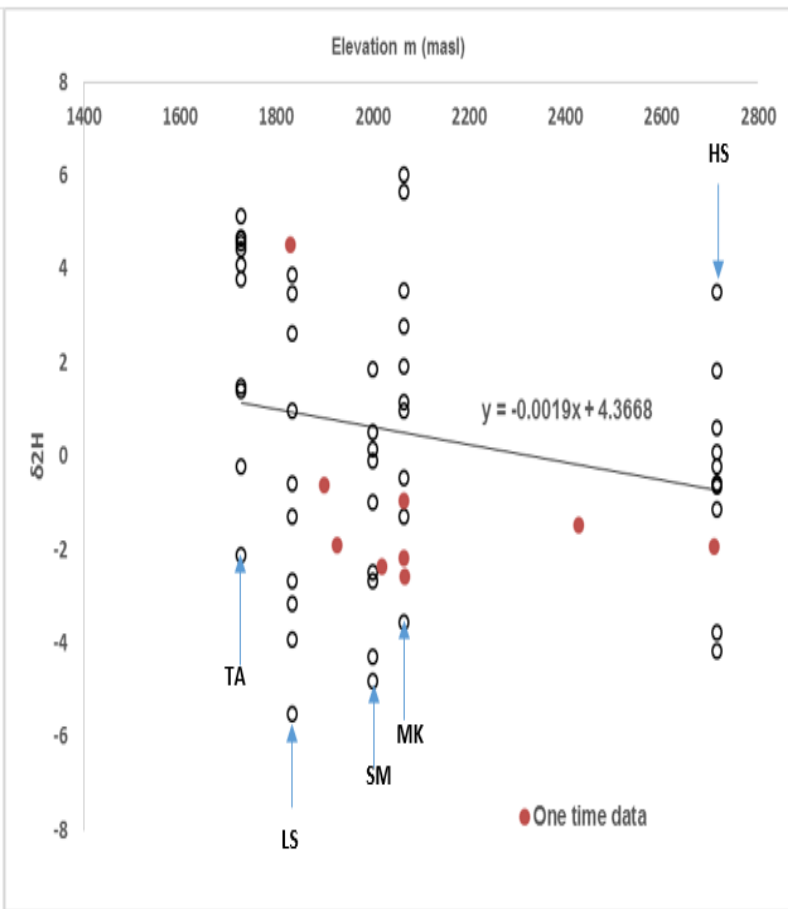
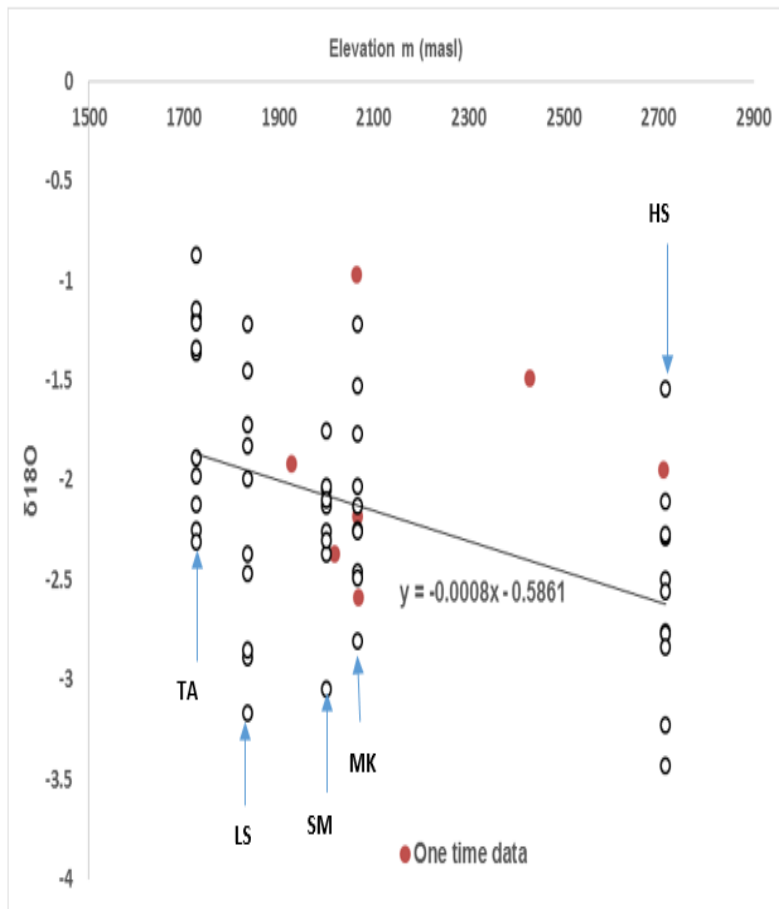
Isotopic composition of the waters of spring and adjacent boreholes

A scatter plot of values of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ of water samples from different sources shows that most of the spring and groundwater data plot close to the LMWL

The data collected from springs and groundwaters have shown that the value of $\delta^{18}\text{O}$ ranges from -3.43‰ to -0.87‰ and $\delta^2\text{H}$ from -10.58‰ to +6.03‰.



Elevation versus stable isotopes

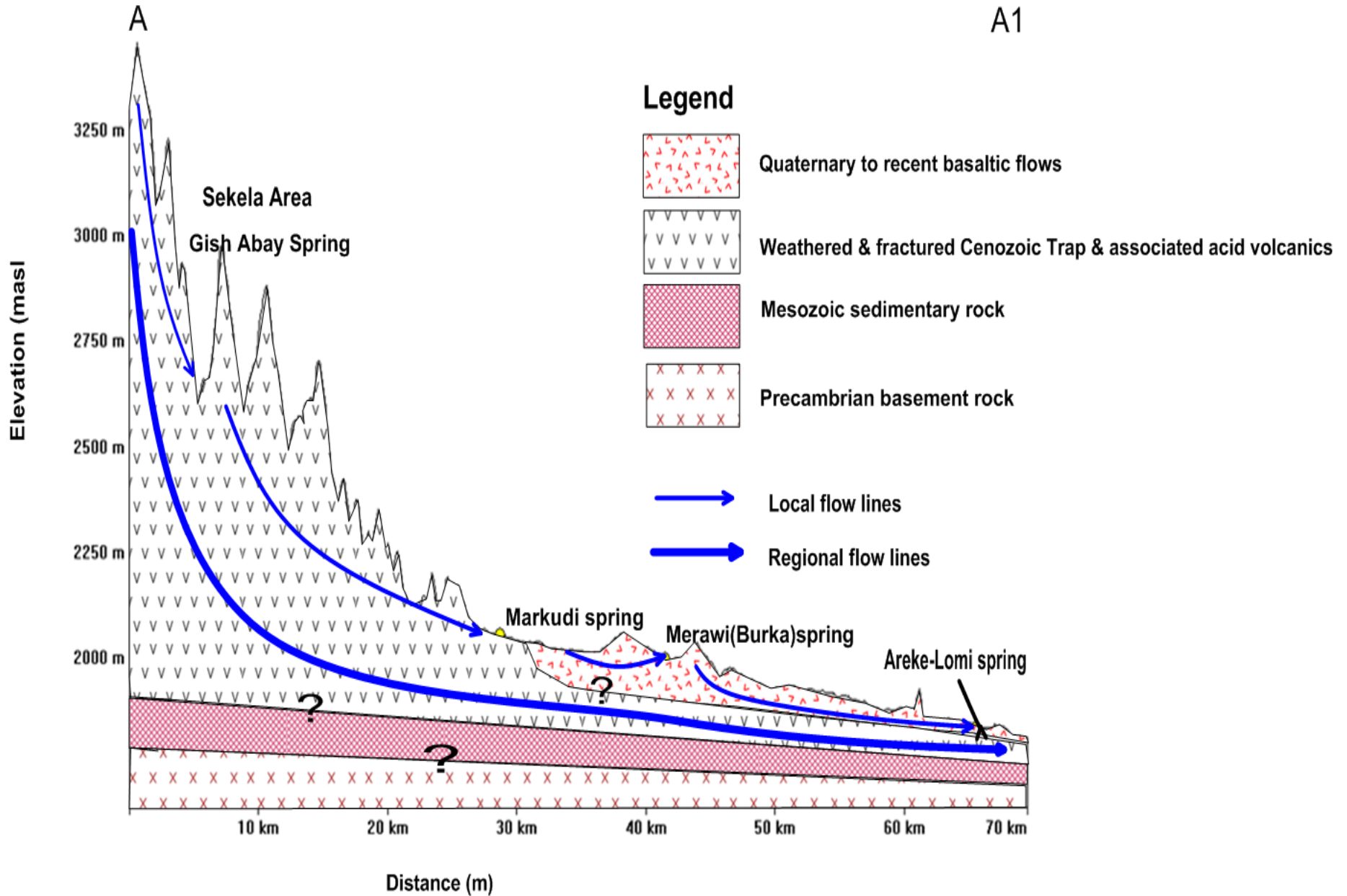


the best fit trend line was calculated
 $\delta^{18}\text{O} = -0.0008 Z - 0.5861$

best fit trend line equation
 $\delta^2\text{H} = -0.0019 Z + 4.3668$

Groundwater flow system and sources of recharge to the springs

- The hydrochemical composition of springs and some groundwater wells is Ca-HCO₃.
 - ❖ This indicates that recharge to springs and groundwater is from the shallow aquifer system with short residence times without further chemical evolution.
- The isotopic signature of most of the samples from boreholes and springs falls within the range of rainwater isotopic composition at Addis Ababa.
 - ❖ This indicates that the sources of recharge for the shallow wells and springs are from modern precipitation



Conclusion

- The springs and most of the well waters sampled for chemical analysis show that the chemistry of the water is categorized into one type of water, Ca-HCO₃.
- The chemical and isotopic composition of spring and borehole water, except some, shows limited water-rock interaction, shallow groundwater circulation, superficial flow and low residence time.
- The isotopic signature of most of the samples from boreholes and springs falls within the range of rainwater isotopic composition.
 - ❖ This indicates that the sources of recharge for the shallow wells and springs are from modern precipitation. Their isotopic composition relates with elevation of the springs, and together with their limited chemical evolution,
 - this shows that they are locally recharged rather than regionally

Thank you