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Water Chemistry and Carbon Flux in Storm Event in a Karst River

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Outline

1. Introduction

2. Site description

3. Methods

4. Discussion

5. summary

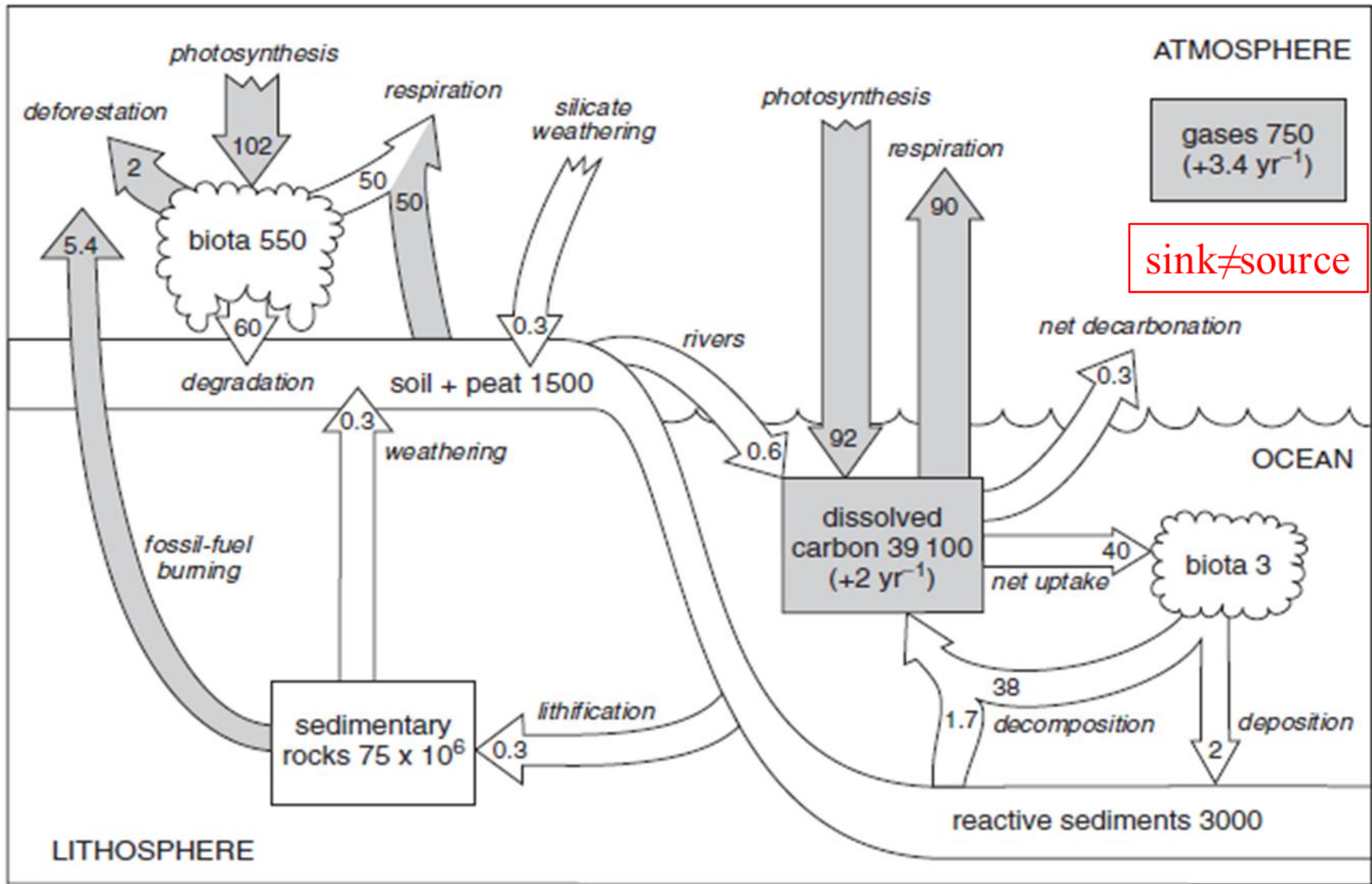


Fig. 6.32 The economic and social development of China is experiencing a booming period, statistics shows that CO₂ emission will reach approximately 127 PgC (exceeding the estimation by 17 PgC), which is leading to a huge CO₂ reduction pressure

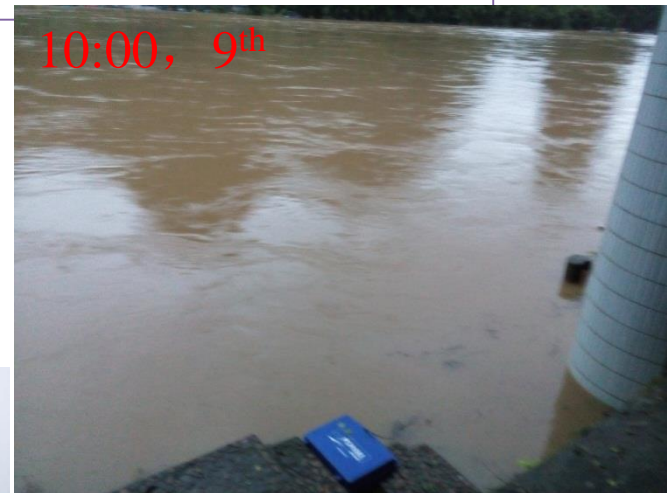
01 - Yangshuo transection

Measurements – surface layer

Sampling(two hour/time)

Automatic monitoring (15min/time)

10:00, 9th



Period

May 7th to 9th, 2015
50 hours in total

14:00, 7th



16:00, 8th



02 - Fuli transection



**Measurements: surface layer and
– 2m layer**

Sampling(two hour/time)

Automatic monitoring(15min/time)

Period

May 5th to 7th, 2015
50 hours in total



3. Methods

Field water chemical parameters

Field indexes	pH	15min/time
	Do	15min/time
	Spc	15min/time
	T	15min/time
	HCO ₃ ⁻	2 hour/time
	Ca ²⁺	2 hour/time



Field water chemical parameters



Total rainfall during monitoring period : 42.5(mm)

Total flow during monitoring period : 9.42E+07 (m³/s)

Laboratory analysis

- Main ions

K^+ 、 Na^+ 、 Ca^{2+} 、 Mg^{2+} 、 SiO_2

SO_4^{2-} 、 NO_3^- 、 HCO_3^- 、 Cl^- 、 F^-





Calculation

$$C_{SF} = \sum_{t1}^{t2} (([HCO_3^-]_{car} + [HCO_3^-]_{sil} + [HCO_3^-]_{acid}) \times Q)$$

- C_{SF} :the carbon flux during the monitoring period(tCO_2)
- $[HCO_3^-]_{car}$:the HCO_3^- formed by carbonate (tCO_2)
- $[HCO_3^-]_{sil}$:the HCO_3^- formed by silicate weathering (tCO_2)
- $[HCO_3^-]_{acid}$:the HCO_3^- formed by carbonate weathering with the participation of allogenic acid (tCO_2)
- Q :the total flow during the monitoring period.



4. Discussion

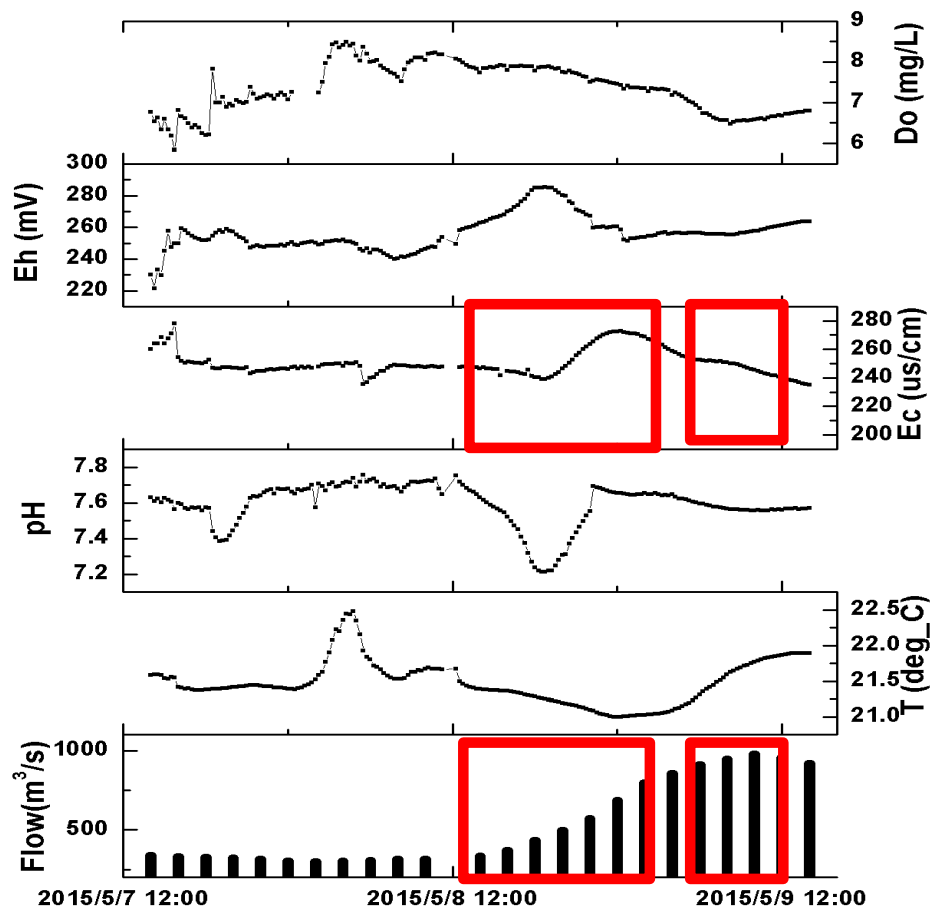
4.1 Water chemical variation

4.2 Major ions variation and

4.3 Inorganic carbon flux

4.4 Inorganic carbon sources

4.1 Water chemical variation



01-Yangshuo transection

(surface layer)

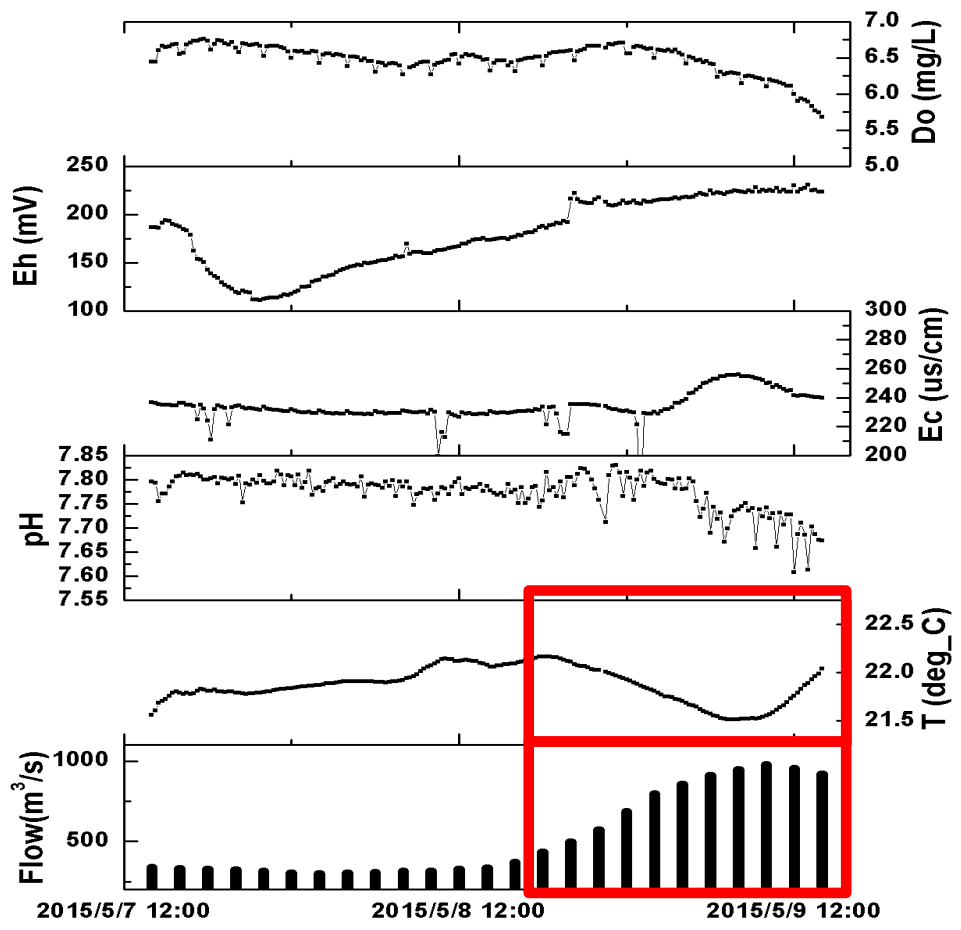
Ec and pH increased as water flow increased, indicating a fast water-rock interaction during the storm event, Ec increased and pH decreased indicating dilution effect was stronger than The piston effect.



Rock weathering was enhanced before the peak and diluted during the peak.

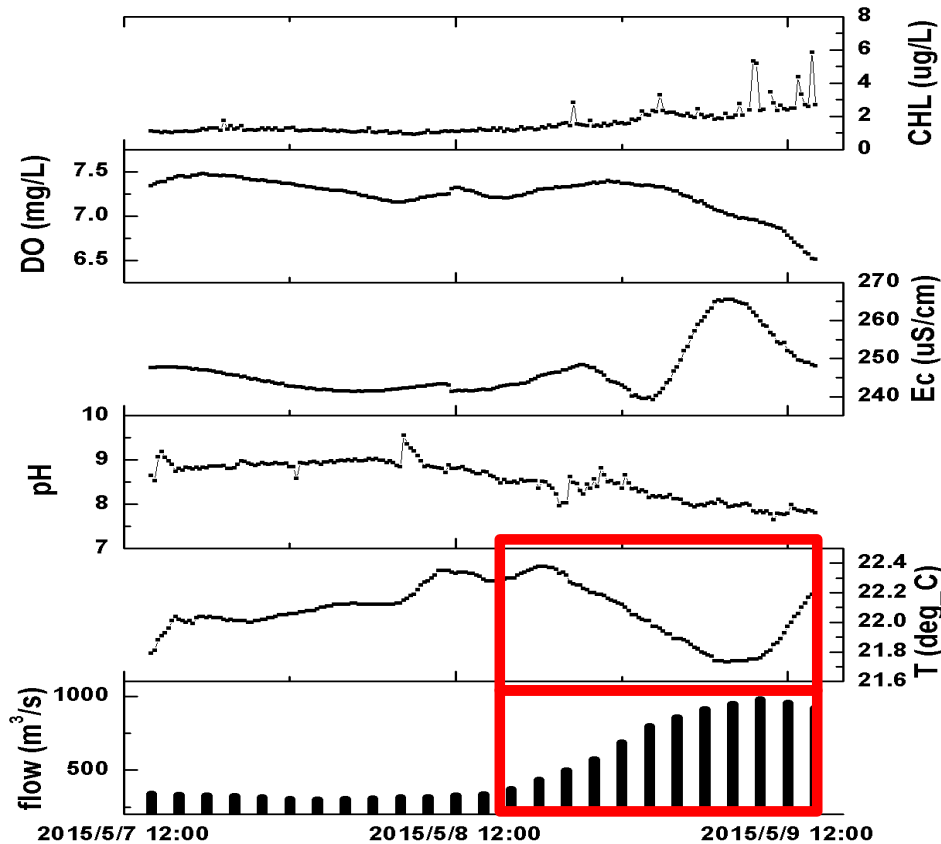


4.1 Water chemical variation



02 - Fuli transection
(surface layer)

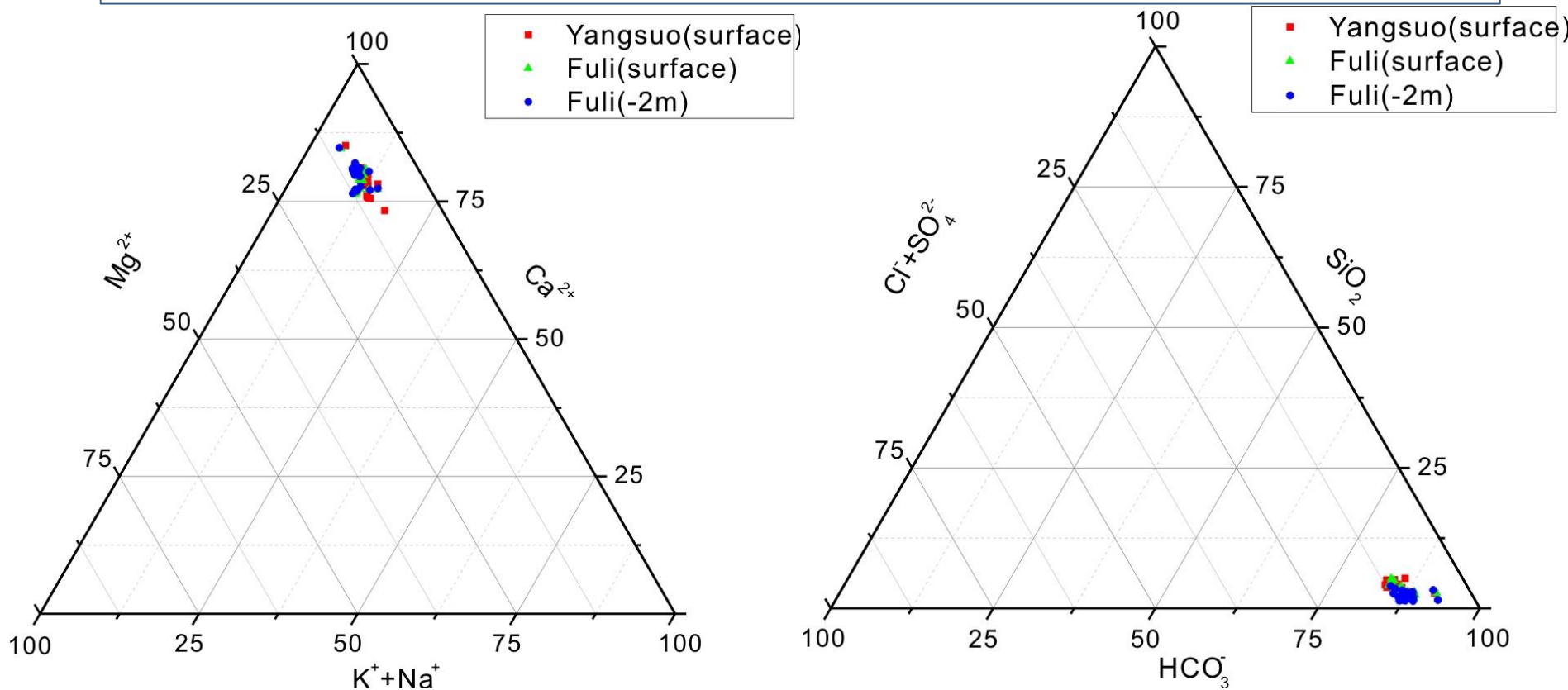
4.1 Water chemical variation



02 - Fuli transection
(-2m layer)

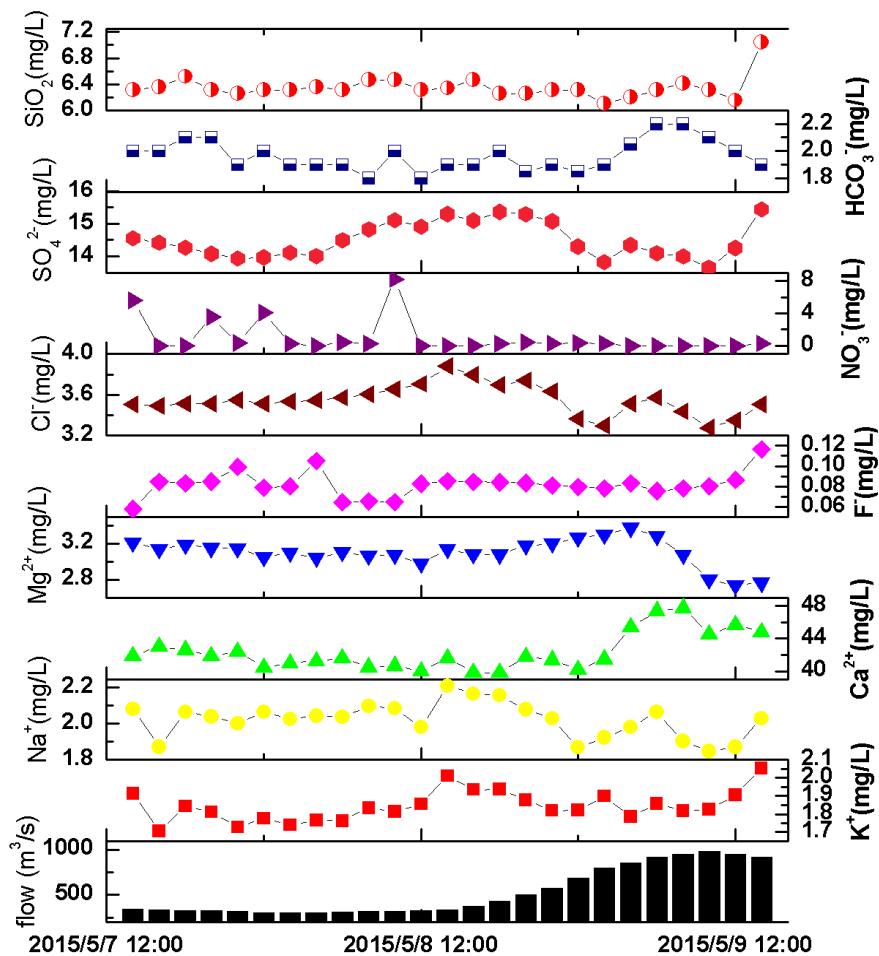
Ec increased, pH decreased, indicating that the water-rock interaction reached its maximum rate under the continuous precipitation, or the rainwater flowed into the river as surface flow

4.2 Major ions variation



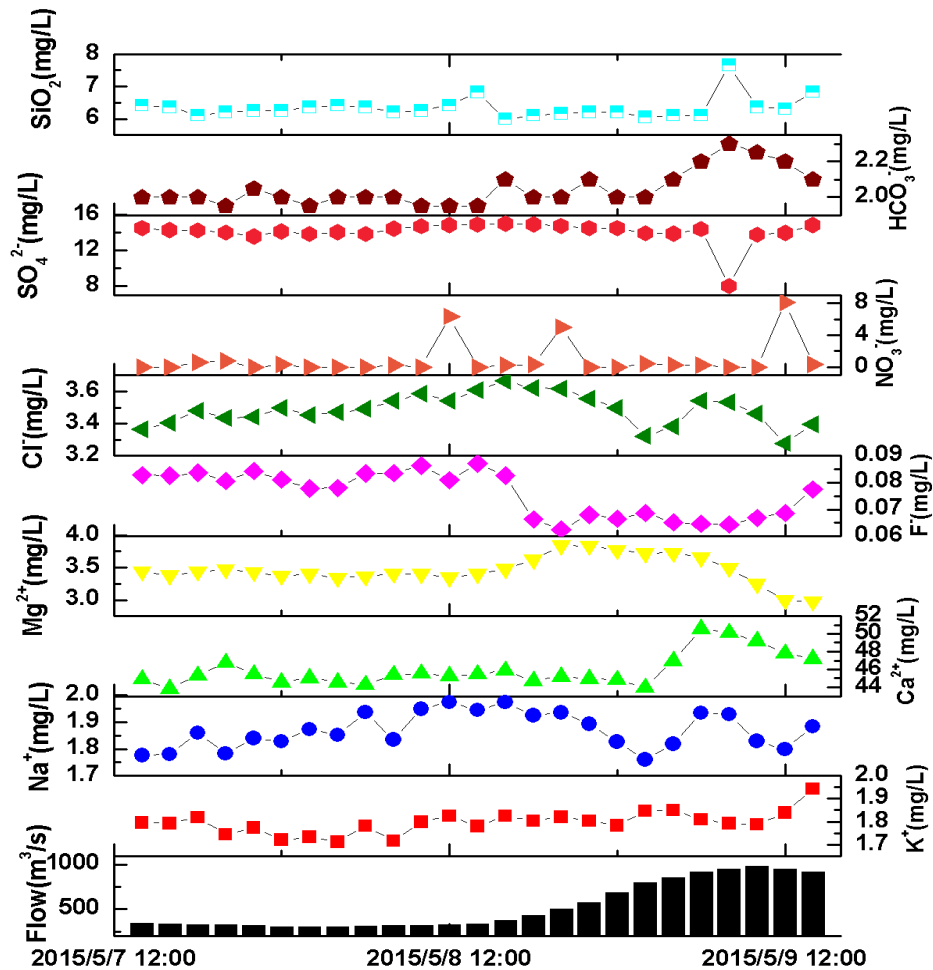
Piper diagram of water samples from three transections during monitoring

4.2 Major ions variation



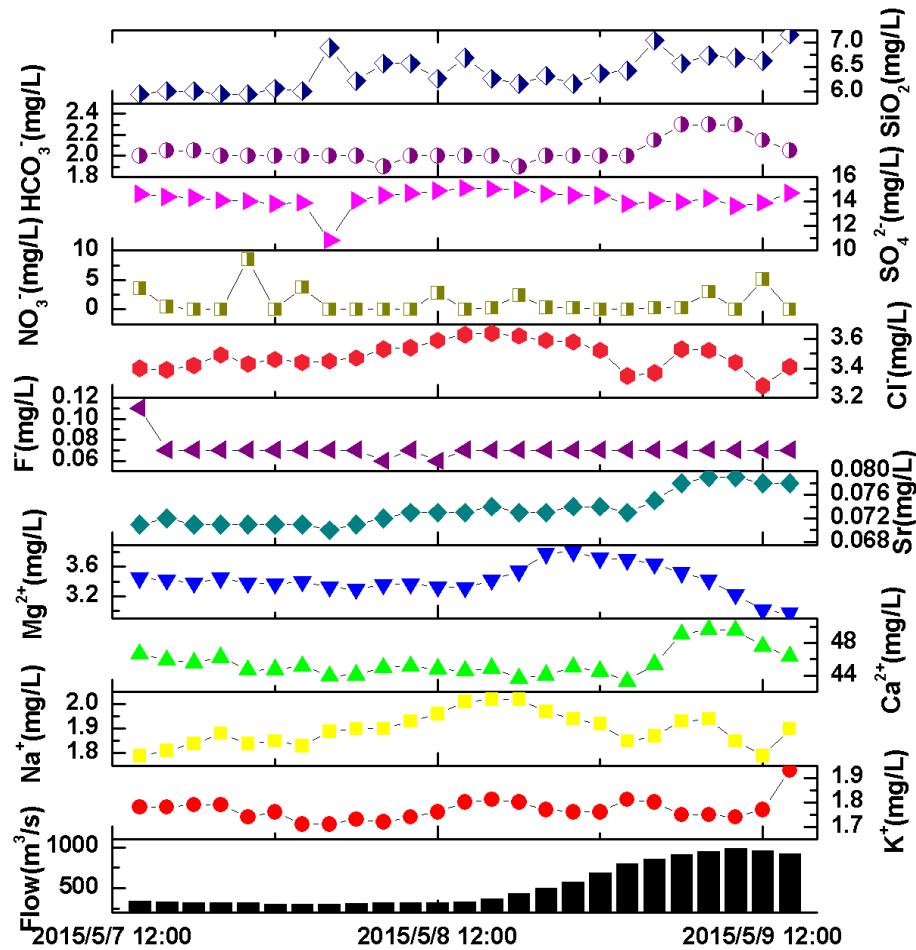
01-Yangshuo transection
(surface layer)

4.2 Major ions variation



02 - Fuli transection
(surface layer)

4.2 Major ions variation



02 - Fuli transection
(-2m layer)



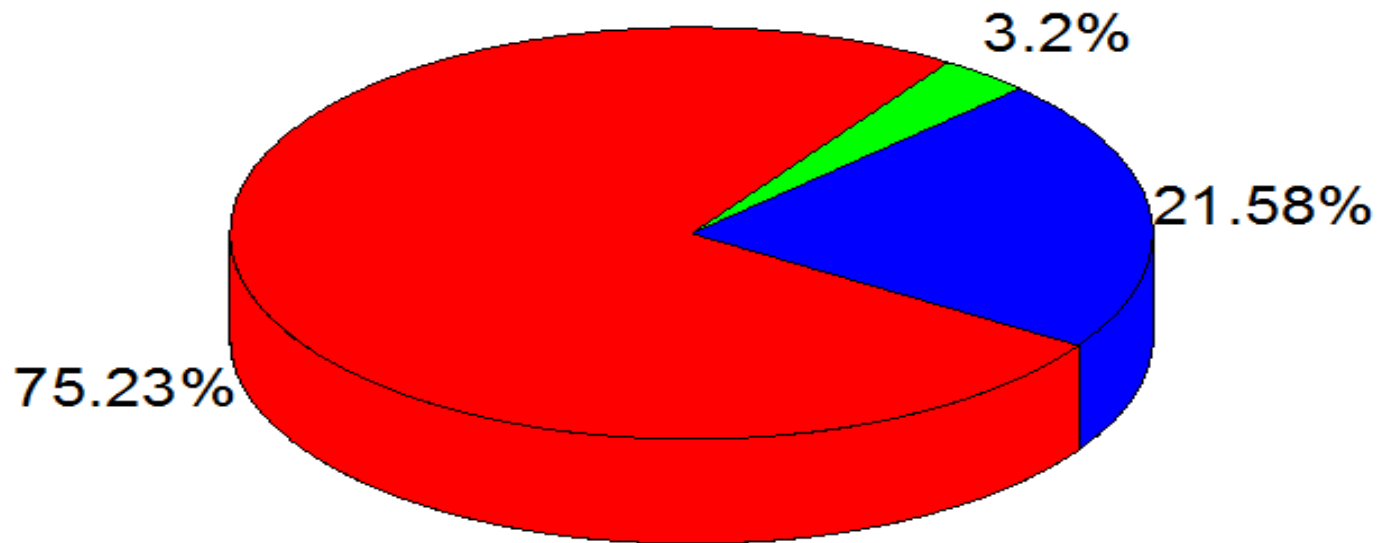
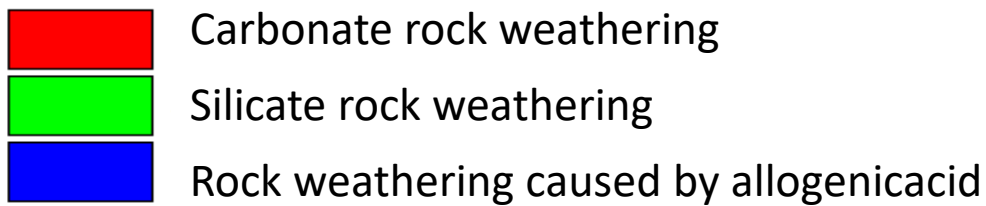
4.3 Inorganic carbon flux

- The total carbon flux increased from upstream (01-Yangshuo) to downstream (02-Fuli)
- Lithology
- Soil

Transection	Carbon flux (tCO ₂)
01-Yangshuo Surface layer	4122.29
02-Fuli Surface layer	4322.14
02-Fuli -2m layer	4324.91

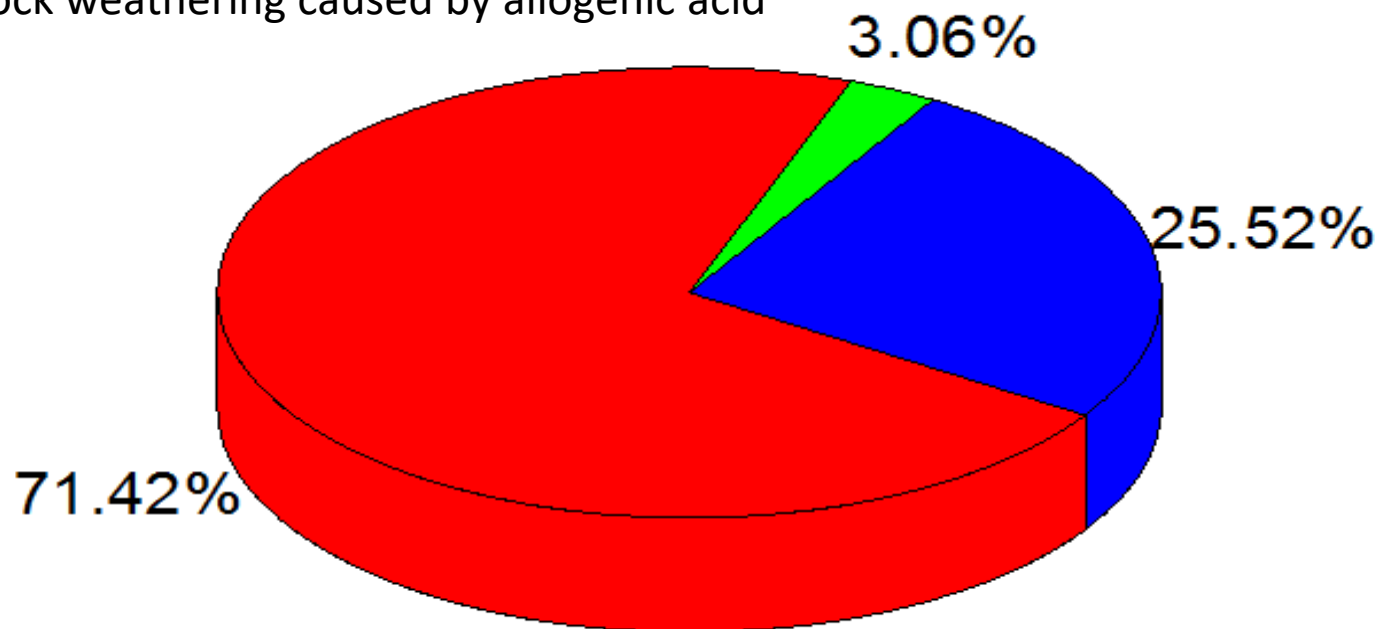
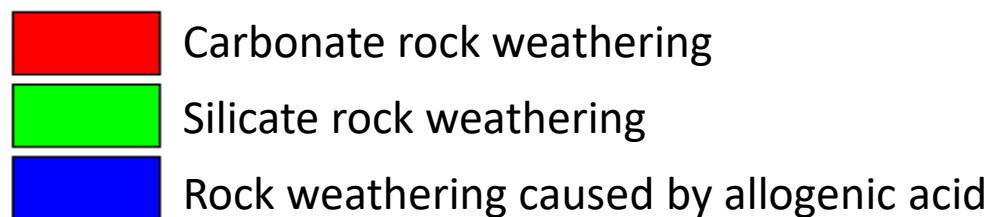
4.4 Inorganic carbon sources

01-Yanshuo transection (Surface layer)



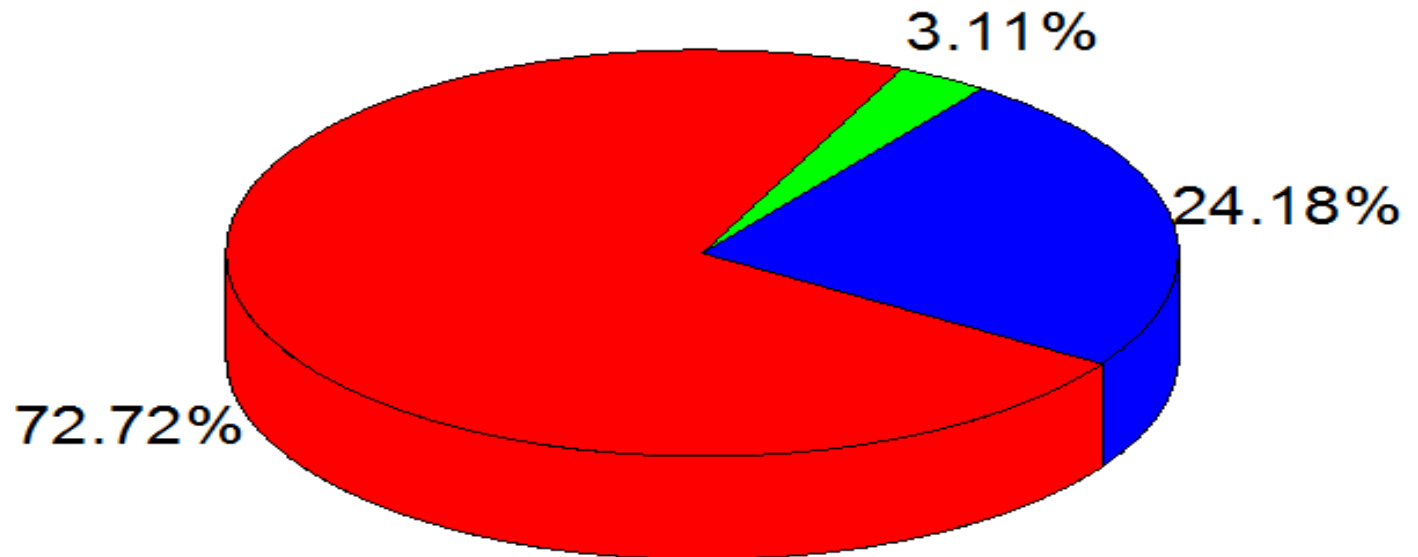
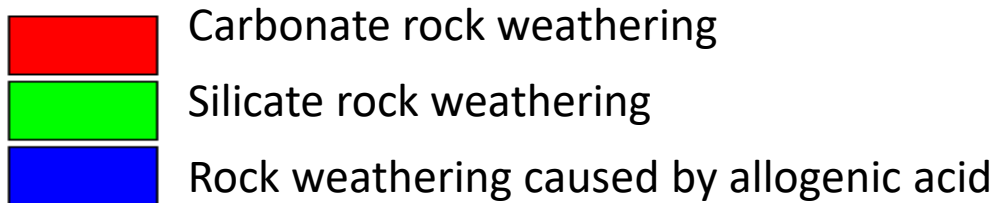
4.4 Inorganic carbon sources

02-Fuli transection (Surface layer)



4.4 Inorganic carbon sources

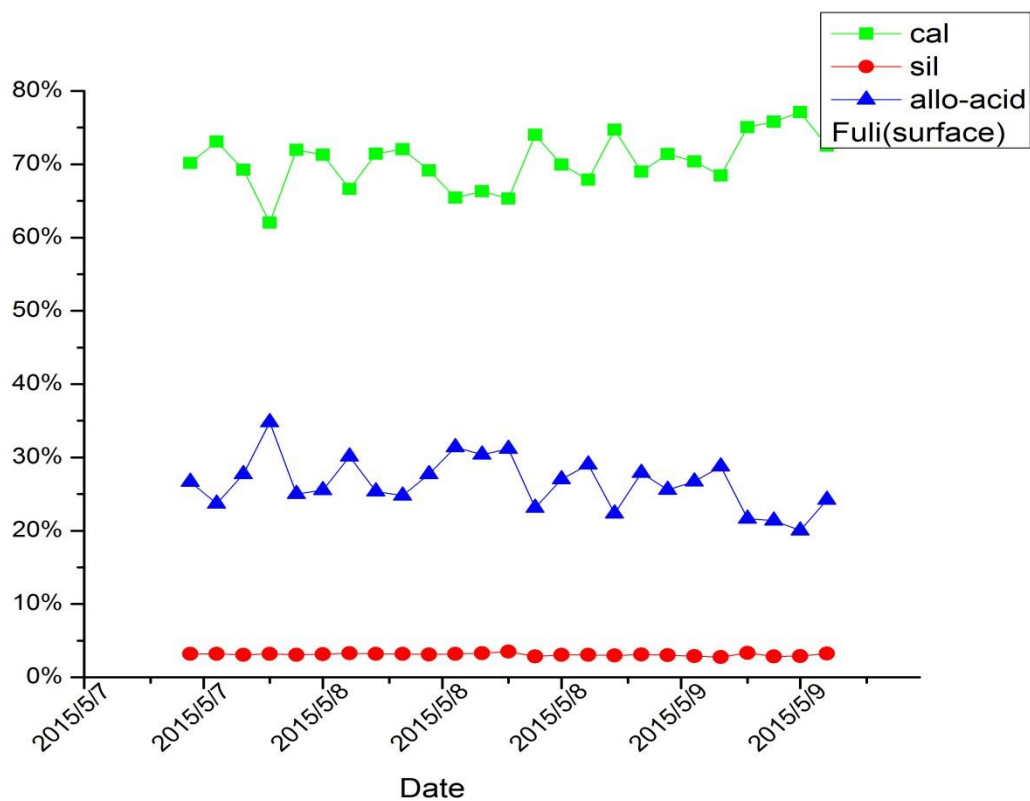
02-Fuli transection (-2m layer)



4.4 Inorganic carbon sources

02-Fuli transection (Surface layer)

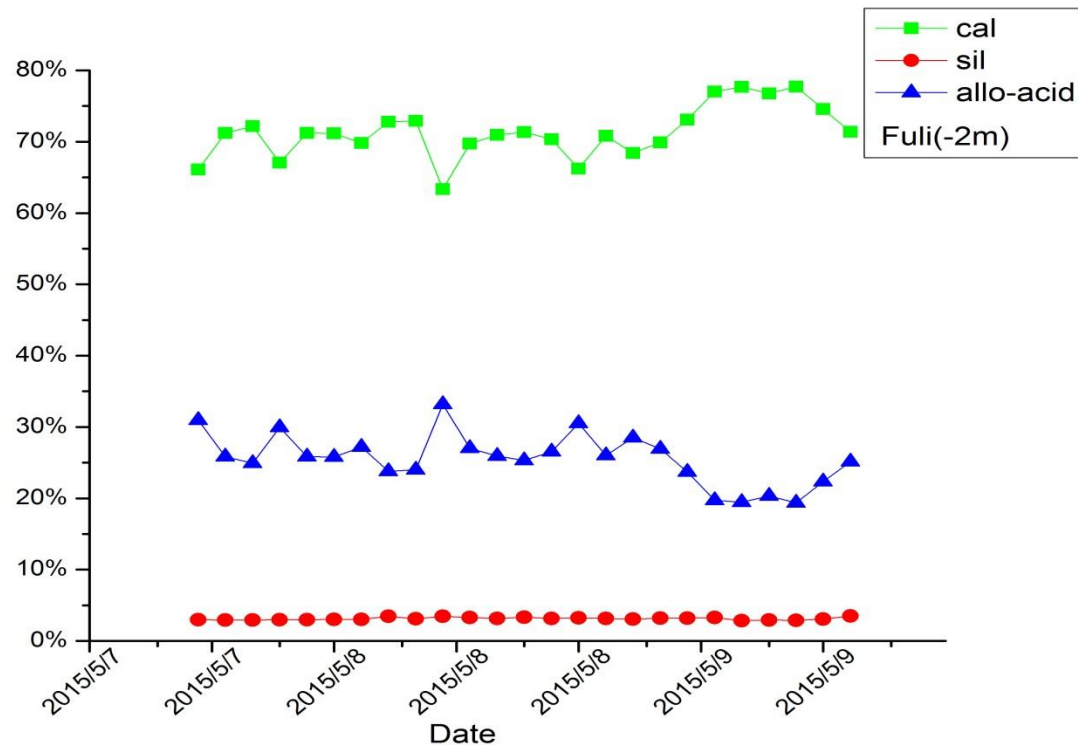
Variations of carbon source contributions against time



4.4 Inorganic carbon sources

02-Fuli transection (-2m)

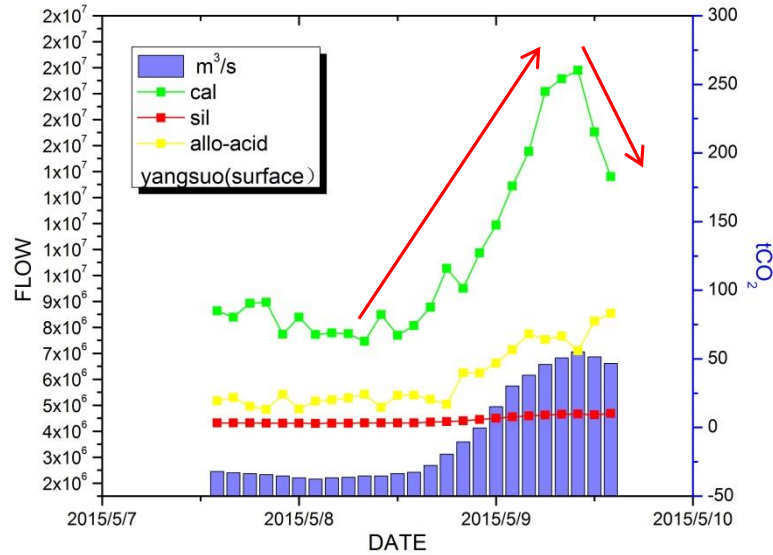
Variations of carbon source contributions against time



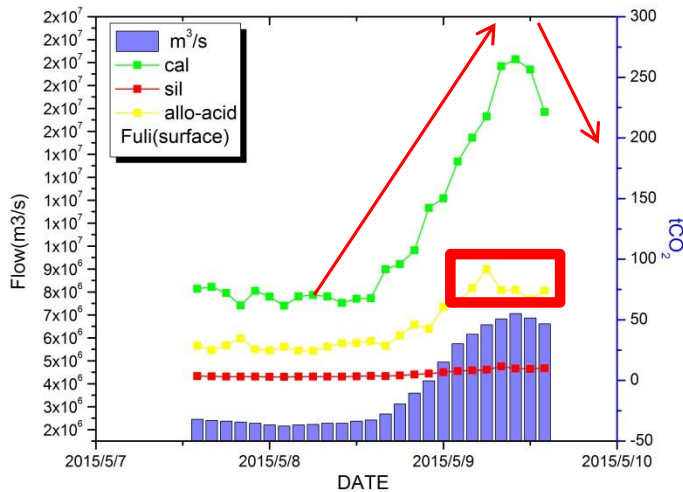
4.4 Inorganic carbon sources



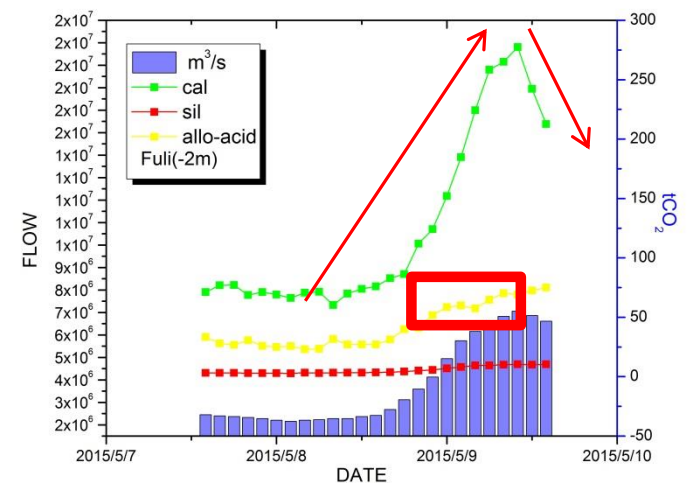
01-Yanshuo transection (Surface layer)



02-Fuli transection (Surface layer)



02-Fuli transection (-2mlayer)





5. Summary

Water chemical

- Hydrochemical facies of the three transections were HCO_3^- -Ca, reflecting lithological control on the Hydrochemical facies
- The variation of Cl^- , F^- , Na^+ , K^+ would come from **varied sources**



5. Summary

Rock weathering

- Rock weathering was enhanced before the peak and diluted during the peak.
- Under the continuously precipitation, the maximum capacity of water-rock interaction was reached, or the rain flowed into river as surface flow.



5. Summary

Carbon flux

- The *CSFs* of Yangshuo, Fuli and Fuli (-2m) transections were 4122.29 tCO₂, 4322.14 tCO₂, and 4324.91 tCO₂
- The percentages of carbonate weathering accounted for 75.23%, 71.42% and 72.72%
- The percentages of silicate weathering were 3.2%, 3.06% and 3.11%
- The proportion of weathering caused by allogenic acid were 21.58%, 25.52% and 24.18%, which caused the distinct carbon loss



Xijiang river
Catchment: 327006km²



Banzhai stream (groundwater-fed)
Catchment: 30km²



Yaji karst experimental site
Catchment: 2km²



Thanks!